Overview of a new Image-Charge Solvation Model (ICSM) for Electrostatic Interactions

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# OUTLINE

### **INTRODUCTION**

- Motivation: Need for a explicit/implicit hybrid model
- Reaction field from solvent using multiple image charges
- Simulation setup for all-atom molecular dynamics

### **RESULTS**

- Properties of bulk water (0 ionic strength)
- Ion solvation (Na<sup>+</sup>, Na<sup>+</sup>-Cl<sup>-</sup>, Na<sup>+</sup>-Na<sup>+</sup>, Cl<sup>-</sup>-Cl<sup>-</sup>)
- Improved image charge formulas

### CONCLUSIONS

Software implementation for TINKER

# Introduction

- Electrostatic interactions play an essential role in the stability and function of biopolymers such as proteins.
- Different computational methods offer different types of accuracy versus efficiency tradeoffs.

#### Two commonly employed types of methods:

- (explicit) All-atom model using periodic boundary conditions.
- (implicit) Continuum solvent model that treats solvent as a dielectric medium.

# **Explicit Solvent Models**

In all-atom simulations, the solvent is treated as a collection of molecules/atoms.

- Strengths: High accuracy when using PBC.
- Weaknesses: Need large unit cells that are computationally intensive for biologically relevant timescales.



# **Implicit Solvent Models**

The solvent is represented as a dielectric continuum and the biopolymer is placed in a low dielectric medium.

- Strengths: High efficiency and no artifacts from periodicity.
- Weaknesses: The information lost about molecular interactions at the molecular surface causes artifacts.



# **Explicit/Implicit Hybrid Model**

The polymer and solvent inside the sphere are treated explicitly. Outside, the solvent is treated as a continuum dielectric medium.

- Strengths: Spherical geometry allows Poisson Boltzmann Eq. to be solved semi-analytically with no artifacts from periodicity.
- Weaknesses: Artifacts are caused by the discontinuous dielectric and molecular interactions at the spherical boundary, and there still remains a lot of explicit solvent.



- How close can we safely come to the spherical surface?

# **Reaction field from Poisson Equation**

The total potential  $\Phi$  inside the sphere equals the sum of the potentials due to all charges inside the sphere, and the potential due to the polarization of the solvent, called the **Reaction Field**.

$$\Phi_{\text{total}} = \Phi_{\text{inside}} + \Phi^{\text{RF}}$$



# **Reaction field by direct expansion**

The total potential  $\Phi$  inside the sphere equals the sum of the potentials due to all charges inside the sphere, and the potential due to the polarization of the solvent, called the **Reaction Field**.

$$\Phi_{\text{total}} = \Phi_{\text{inside}} + \Phi^{\text{RF}}$$

#### For 1 source charge along the x-axis:

Due to azimuthal symmetry, the reaction potential  $\Phi(r,\theta)$  can be expressed in terms of Legendre polynomials of  $\cos(\theta)$ :



$$\Phi^{RF}(r) = \sum_{n=0}^{\infty} B_n r^n P_n(\cos\theta), r \le a ,$$

$$B_n = \frac{q}{4\pi\varepsilon_i} \cdot \frac{r_s^n}{a^{2n+1}} \cdot \gamma \cdot \left(1 + \frac{1+\gamma}{1-\gamma+2n}\right), \quad n \ge 0, \quad \gamma = \frac{\varepsilon_i - \varepsilon_o}{\varepsilon_i + \varepsilon_o} .$$





*Numerical Problem*: Replacing the integral over the line charge density with Jacobi - Gauss Quadrature involving M discrete points yields:



*Physical interpretation:* The reaction potential can be approximated by a Kelvin image charge and a finite number of additional point charges.

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**Example: Protein 1BNI plus water molecules (not shown) and their images** 

- Purple dots: Partial charges on the protein
- Red dots: Image charges associated with the protein
- Blue dots: Image charges associated with explicit water molecules in the sphere

# **Setup for Molecular Dynamics Simulation**



### FEATURES

### Simulation Box

truncated octahedron with periodic boundaries for short range forces.

**Long range electrostatics** Fast Multipole Method (FMM)

**Region I** production volume

**Region II** imaged water for Region III

### Region III

buffer region required

### **Bulk water simulation**

Require structural and dynamic properties of water in the production volume to agree with PME calculations.

Y. Lin, A. Baumketner, S. Deng, Z. Xu, D. Jacobs, W. Cai, J. Chem. Phys. 131:154103 (2009)



Bulk properties of water are observed Example: Density of water

The density of water measured at 11 different bins along the diagonal of the TO-box is uniform in general, except for too small of a buffer region.

Finite size effects are small, but show up for small box size (L=30 Ang.) with small buffer region (4 Ang. or less).

#### Standard deviation in relative density

	<i>τ</i> =2 Å	<i>τ</i> =4 Å	τ=6 Å	<i>τ</i> =8 Å
<i>L</i> =30 Å	0.056	0.011	0.003	0.002
L=45 Å	0.060	0.007	0.002	0.002
<i>L</i> =60 Å	0.055	0.009	0.002	0.003





The plateau values are approximately equal to the expected bulk value.

Dielectric constant calculated over a spherical ball of radius R.



Only one image charge (the Kelvin image) is sufficient to reproduce bulk water properties. No image charge creates unphysical results.

### **Results of Molecular Dynamics Simulation** Electrostatic force distributions





#### Free energy of Na<sup>+</sup> solvation: Using a single ion

# Estimates using the ICSM agree with previously published accurate estimates

TABLE IV. Comparison of charging free energy  $\Delta G_{cg}^s$  of sodium ion computed in this work by the ICSM and SSBP model (Ref. 70) with the values reported for systems of comparable dimensions by PME, straight cut-off method and finite-drop model (Ref. 49).



Method/model	$\Delta G_{cg}^{s}$ (kJ/mol)	
ICSM (30 Å box)	-445	
PME with <i>P</i> -type correction (Ref. 49) (31.3 Å box)	-420	Not accounting
Straight cut-off with $R_c = 14$ Å (Ref. 49) (31.3 Å box)	-442	→ for finite size
Finite droplet (Ref. 49) (18 Å radius)	-429	effects
SSBP (15 Å radius)	-388	
Reference value (Ref. 49)	-480	—— best estimate

<sup>49</sup>M.A. Kastenholz and P.H. Huenenberger. J. Chem. Phys. 124, 224501 (2006)

Y. Lin, A. Baumketner, W. Song, S. Deng, D. Jacobs, W. Cai, J. Chem. Phys. 134:044105 (2011)

#### Free energy of Na<sup>+</sup> solvation: Using a single ion

# Estimates using the ICSM agree with previously published accurate estimates

TABLE I. Electrostatic solvation energy  $\Delta G_{cg}$  (in kJ/mol) of a sodium ion in water obtained in this work in simulation boxes of different sizes L. Finitesize corrections  $\Delta G_c$  are evaluated separately from the charging free energy  $\Delta G_{cg}^s$  computed directly in the simulations (explained in the text).

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L ( Å)	$\Delta G^s_{ m cg}$	$\Delta G_c$	$\Delta G_{ m cg}$
30	-445	-64	-509
45	-457	-33	-490
60	-465	-20	-485

Accounting for finite size effects tends to -480 in a systematic way.

Y. Lin, A. Baumketner, W. Song, S. Deng, D. Jacobs, W. Cai, J. Chem. Phys. 134:044105 (2011)

Free energy of Na<sup>+</sup> solvation: Using a single ion

Estimates using the ICSM agree with previously published accurate estimates

TABLE II. Charging free energy  $\Delta G_{cg}^s$  computed for different ion locations in the 60 Å simulation box.

Distance to origin (Å)	$\Delta G_{cg}^{s}$ (kJ/mol)
0	-465
6.0	-468
12.0	-467
18.0	-468
24.0	-459

No systematic dependence on position of ion in the TO-box.

Y. Lin, A. Baumketner, W. Song, S. Deng, D. Jacobs, W. Cai, J. Chem. Phys. 134:044105 (2011)





Potential of Mean Force (PMF) for ion pairs

Results using ICSM agree with previously published accurate results.

Y. Lin, et. al., J. Chem. Phys. 134:044105 (2011)

# **Improved image charge formulas**

#### **Build in a transition layer?**





# **Improved image charge formulas**

### preliminary results

#### See poster by Wei Song

Optimized image charges for reaction field calculations

Ratio of optimized image charge to that of the DDM as a function of source charge location.



# Conclusions

- Our results on bulk water simulations and ion charging free energies suggest the **Image Charge Solvation Model (ICSM)** provides a high accuracy calculation for the reaction field while minimizing finite size effects during the molecular dynamics simulation.
- The reaction field is weak compared to direct local interactions with about a 1% effect on atomic forces, but it is critical to get local structure of water correct due to torques on molecular orientations where there is about a 20% effect.
- The ICSM can be generalized to ionic solutions. More will be said about image charge methods by Wei Cai in his talk on *Image Approximations to Reaction Fields in Inhomogeneous Media* this afternoon.
- The ICSM is implemented in TINKER. Preliminary benchmarks show that the crossover point for ICSM to be faster than PME is a TO-box size of 80 Angstroms corresponding to about 25,000 atoms.